

1.

a. The Nernst equation for ion [i] is given by

$$E_i = \frac{RT}{zF} \ln \left( \frac{C_{out,i}}{C_{in,i}} \right)$$

where  $z$  - valence charge,  $C_{out}$  the ion concentration outside the cell,  $C_{in}$  the ion concentration inside the cell,  $R$  - thermodynamic gas constant,  $F$  - Faraday constant and  $T$  - temperature in Kelvin.

Given  $R = 8.31447$  [J/mol·K],  $T = 273 + 25$  [K] and  $F = 9.648534 \cdot 10^4$  [C/mol] then  $RT/F = 0.0257$  [V] or 25.7 [mV].

Using the Nernst equation with the given values of the inner/external concentrations as well as the valence charge results in  $E_1 = 23$ ,  $E_2 = -72$  and  $E_3 = -12$  [mV].

b. A decrease in  $T$  results in the coefficient  $RT/F$  being smaller. Therefore a decrease in  $T$  results in a decrease in  $E_1$  and an increase in  $E_2$  and  $E_3$ .

2.

a. The solution of  $\dot{C} = -0.8C$  is given by  $C(t) = C_0 e^{-0.8t}$ . Solve  $0.4C_0 = C_0 e^{-0.8T}$  for  $T$  to get the time.

$$0.4C_0 = C_0 e^{-0.8T}$$

$$0.4 = e^{-0.8T}$$

$$\ln(0.4) = -0.8T$$

$$T = \frac{\ln(0.4)}{-0.8} = 1.14 \text{hours} \approx 69 \text{minutes.}$$

b. The kinetics of the system is given by

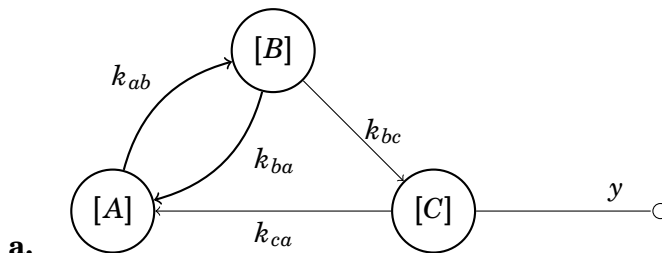
$$\dot{C} = -0.8C + \frac{u}{V}$$

The transfer function of the system is

$$C(s) = \frac{1}{V} \frac{1}{s + 0.8}$$

The static gain is  $C(0) = 1/(0.8V)$ . And the constant dose that achieves the steady state is given by  $u = C_{opt}/C(0)$ .

3.



or use  $(\alpha, V_a)$ , and similarly, in each compartment.

b. Let  $x = [a \ b \ c]^T$ . Then the system can be written as

$$\dot{x} = \begin{bmatrix} -k_{ab} & k_{ba} & k_{ca} \\ k_{ab} & -(k_{ba} + k_{bc}) & 0 \\ 0 & k_{bc} & -k_{ca} \end{bmatrix} x$$

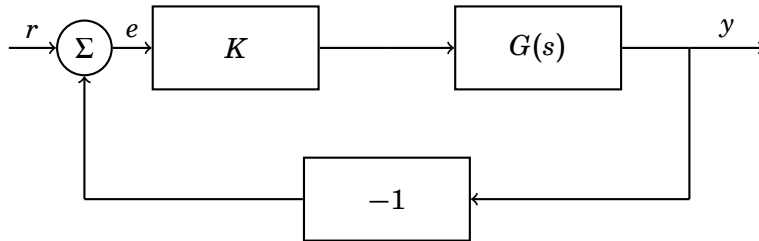
$$y = [0 \ 0 \ 1/V_c] x.$$

c. It is easy to check that  $\dot{a} + \dot{b} + \dot{c} = 0$ . Thus  $a(t) + b(t) + c(t) = q$ , where  $q$  is some constant. Using the initial conditions we find that  $q = p$ .

4.

a. From the Nyquist plot it is clear that the amplitude margin is  $A_m = 1/2$ , and thus the largest  $K$  is  $1/2$ .

b.



c. Inserting  $u = -Lx + l_r r$  gives  $\dot{x} = (A - BL)X + Bl_r r$ . The Laplace transform then gives  $X(s) = (sI - A + BL)^{-1} Bl_r R(S)$ . Thus the transfer function can then be found,

$$Y(s) = \underbrace{C(sI - A + BL)^{-1} Bl_r}_{G(S)} R(S)$$

d. Calculations gives

$$|sI - A + BL| = \det \begin{bmatrix} s - 1 + l_1 & l_2 \\ -2 & s - 1 \end{bmatrix} = s^2 + (l_1 - 2)s + 1 - l_1 + 2l_2$$

This is to be compared with  $(s + 1)^2 = s^2 + 2s + 1$ . Identification of coefficients gives  $l_1 = 4$  and  $l_2 = 2$ .

5.

a.  $\dot{x} = 0$  for  $x = 0$  or  $y = \alpha/\beta$  assuming  $x \neq 0$ . Likewise for  $\dot{y}$  we have that  $y = 0$  or  $x = \gamma/\delta$  assuming  $y \neq 0$ . Thus the two stationary points that makes  $\dot{x} = 0$  and  $\dot{y} = 0$  is  $(0, 0)$  and  $(\gamma/\delta, \alpha/\beta)$ .

b. Let

$$f = \begin{bmatrix} \dot{x} \\ \dot{y} \end{bmatrix}.$$

Then

$$\frac{\partial f}{\partial x}(x, y) = \begin{bmatrix} \alpha - \beta y \\ \delta y \end{bmatrix}, \quad \frac{\partial f}{\partial y}(x, y) = \begin{bmatrix} -\beta x \\ \delta x - \gamma \end{bmatrix}.$$

In the point of interest we have

$$\frac{\partial f}{\partial x}(x^0, y^0) = \begin{bmatrix} 0 \\ \delta \frac{\alpha}{\beta} \end{bmatrix}, \quad \frac{\partial f}{\partial y}(x^0, y^0) = \begin{bmatrix} -\beta \frac{\gamma}{\delta} \\ 0 \end{bmatrix}.$$

Let  $\Delta x = x - x^0$  and  $\Delta y = y - y^0$ . Then we have that

$$\begin{bmatrix} \Delta \dot{x} \\ \Delta \dot{y} \end{bmatrix} = \begin{bmatrix} 0 & -\beta \frac{\gamma}{\delta} \\ \delta \frac{\alpha}{\beta} & 0 \end{bmatrix} \begin{bmatrix} \Delta x \\ \Delta y \end{bmatrix}.$$

- c.** The poles of the systems are given by the solution to  $s^2 + \alpha\gamma$ . Thus the poles are given by  $\pm i\sqrt{\alpha\gamma}$ . The poles are on the imaginary axis and the linearized system is marginally stable.

**6.**

- a.** Assuming steady state, the output is given by  $y(t) = |G(3i)| \sin(3\omega t + \arg(G(3i)))$ .

$$|G(3i)| = \frac{2}{\sqrt{9+9}} = \frac{2}{\sqrt{18}} = \frac{\sqrt{2}}{3}$$

$$\arg(G(3i)) = -\arctan(1) = -\pi/4$$

It is also possible to use the method in the next subproblem.

- b.** We have that  $Y(s) = G(s)U(s) = 2/(s(s+3))$  Using the collection of formulae we find that

$$y(t) = 2/3 \cdot (1 - e^{-3t})$$

- c.** The output of the first system can be found by

$$y_1(t) = \int_0^t h(t-\tau)u_2(\tau) d\tau,$$

where  $h$  is the impulse response of the linear system. Similarly for the second system

$$\begin{aligned} y_2(t) &= \int_0^t h(t-\tau)u_2(\tau) d\tau = \int_0^t h(t-\tau)2u_1(\tau) d\tau \\ &= 2 \int_0^t h(t-\tau)u_1(\tau) d\tau = 2y_1(t) \end{aligned}$$

System B can not be linear as the responses from two different step sizes can not be related only by a positive scaling factor.

- 7.** a - ii, b - iv, c- i, d-iii.

The step response in (a) have static gain 0.5.

The step response in (b) have a time delay.

The step response in (c) have static gain 1, but no time delay.

The step response in (d) is unstable (and have complex poles).

Tabell 1: Drug data for problem 1

Substrate concentration $[S]$ [units]	Reaction rate $V$ [units/days]
0.5	0.12
1	0.27
2	0.33
4	0.58

8. Rewrite relationship as

$$\frac{1}{V} = \frac{K_m}{V_{max}} \frac{1}{[S]} + \frac{1}{V_{max}}.$$

Now, the parameters  $K_m/V_{max}$  and  $1/V_{max}$  may be estimated as follows: Let the regressor matrix be

$$\Phi = \begin{pmatrix} 1 & 1/[S]_1 \\ 1 & 1/[S]_2 \\ 1 & 1/[S]_3 \\ 1 & 1/[S]_4 \end{pmatrix} = \begin{pmatrix} 1 & 2 \\ 1 & 1 \\ 1 & 0.5 \\ 1 & 0.25 \end{pmatrix}$$

where  $[S]_i$  is the  $i$ -th value of  $[S]$  in Table 1. Denote

$$y = (1/V_1 \quad 1/V_2 \quad 1/V_3 \quad 1/V_4)^T = (8.33 \quad 3.70 \quad 3.03 \quad 1.73)^T$$

where  $V_i$  is the  $i$ -th value of  $V$  in Table 1.

The least-squares solution is then

$$\begin{pmatrix} \frac{1}{\hat{V}_{max}} \\ \frac{\hat{K}_m}{\hat{V}_{max}} \end{pmatrix} = (\Phi^T \Phi)^{-1} \Phi^T y = \begin{pmatrix} 0.77 \\ 3.66 \end{pmatrix}$$

which results in  $\hat{V}_{max} = 1.30$  and  $\hat{K}_m = 4.77$ .

**Good Luck!**